

THEOPHYLLINE DERIVATIVES 5<sup>1</sup>. SELECTIVE ALTERNATIVE REACTIVITY  
OF 7-CHLOROMETHYL-OXAZOLOPURINE TOWARDS POTASSIUM PHTALIMIDE

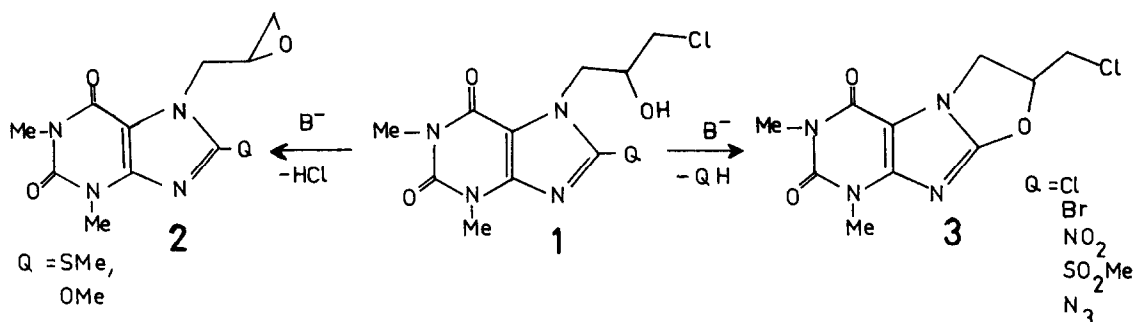
Rudolf Szebeni and Dezső Korbonits

Chinoin Pharmaceutical and Chemical Works LTD

H-1325 Budapest, Pf.110. Hungary

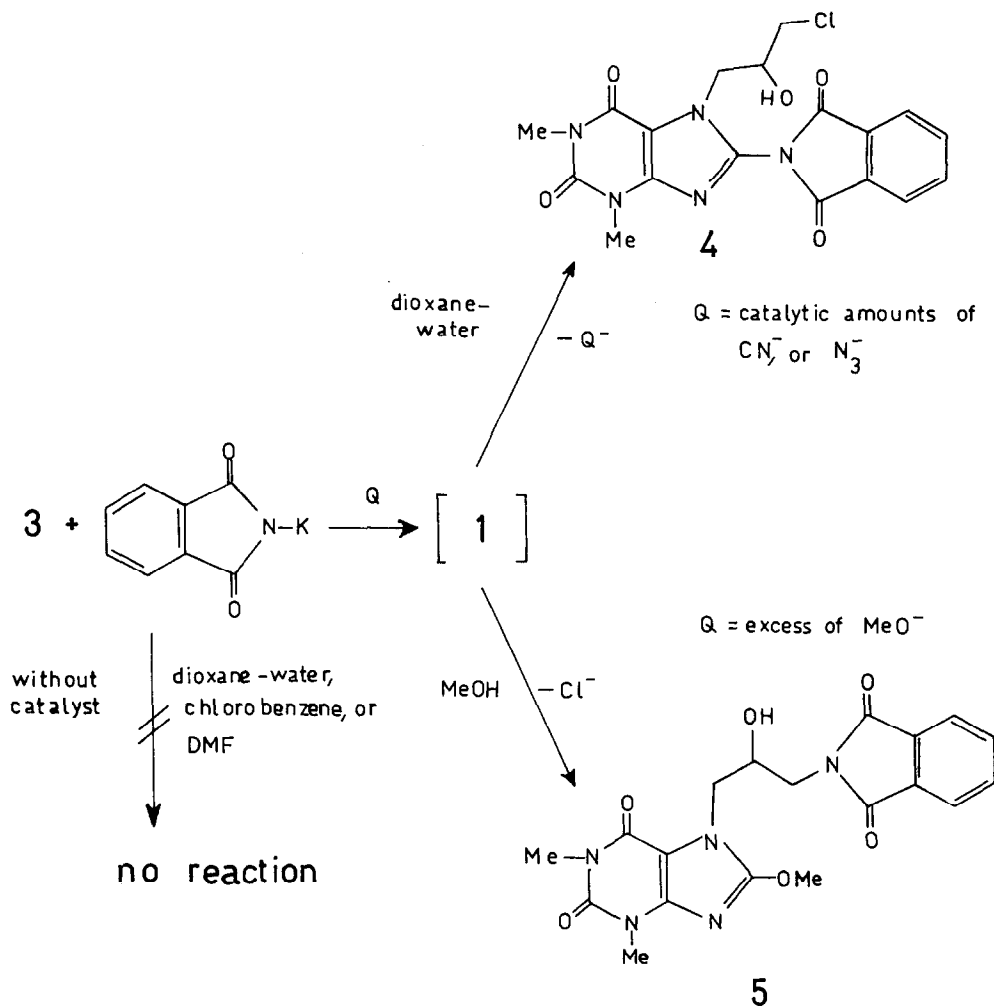
(Received in UK 16 March 1978; accepted for publication 13 April 1978)

Earlier we have discussed correlation between leaving group properties of C<sup>8</sup> substituents in purines 1 and in the epoxypropyl derivatives 2 and the formation of the chloromethyl-oxazolopurines 3 further exchange of chlorine and the cleavage of the C<sup>9</sup>-O<sup>8</sup> bond by various nucleophiles in 3<sup>1,2</sup>. We have established that mobility of chlorine was much restricted by the oxazolidine ring and that of the two electrophilic centers /CH<sub>2</sub> of the side chain, C<sup>9</sup> of the tricycle/ coupled with the opening of the oxazolidine ring, first C<sup>9</sup> was substituted<sup>2</sup>. This was in accord with findings of RYBÁR et al.<sup>3</sup> who reported that with 8-substituted 7-halogenoalkyl-theophyllines deactivation of halogen atoms at the C<sup>7</sup> side chain was due to steric hindrance by the C<sup>8</sup> substituents.



Now we have found that both side chain chlorine and oxazoline ring of 3 are remarkably resistant towards potassium phthalimide as nucleophile, a result which can be mainly attributed to steric hindrance. Earlier we have observed<sup>4</sup> that some small and highly nucleophilic anions, e.g. cyanide, which could be also eliminated, catalysed efficiently attack at the C-O bond of heterocyclic ethers. Accordingly when 3 was reacted with potassium phthalimide in the presence of 0.01 molar eq. of KCN or NaN<sub>3</sub> the phthalimido-purine 4 was obtained in good yield. An increase of molarity of either the catalyst or of potassium phthalimide did not result in further substitution indicating that in 4 chlorine was also deactivated by the bulky phthalimido group<sup>3</sup>.

Transformation 3-4 is presumably initiated by oxazolidine ring opening by  $\text{CN}^-$  or  $\text{N}_3^-$  resulting in 1 /  $\text{Q}=\text{CN}$  or  $\text{N}_3$  /. Steric hindrance is thus released and gives way to substitution of the cyanide or azide group by phthalimide to give a  $\text{C}^8$  substituted theophylline. If a small nucleophile capable to cleave the  $\text{C}^9-\text{O}^8$  bond but not prone to substitution was applied, the deactivating effect of the oxazoline ring was lifted and chlorine at the side chain could be substituted by phthalimide. In fact reaction of 3 with potassium phthalimide in the presence of excess methoxide gave 5.



#### References:

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